

RECELL-BATTERY DESIGN FOR RECYCLING

Project ID: bat467



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Oak Ridge National Laboratory

2021 DOE Vehicle Technologies Office
Annual Merit Review

PROJECT OVERVIEW

Timeline

- Project start: October 2018
- Project end: September 2021
- Percent complete: ~90%

Budget

FY19	\$4,615k
FY20	\$5,150k
FY21	\$4,915k

Barriers

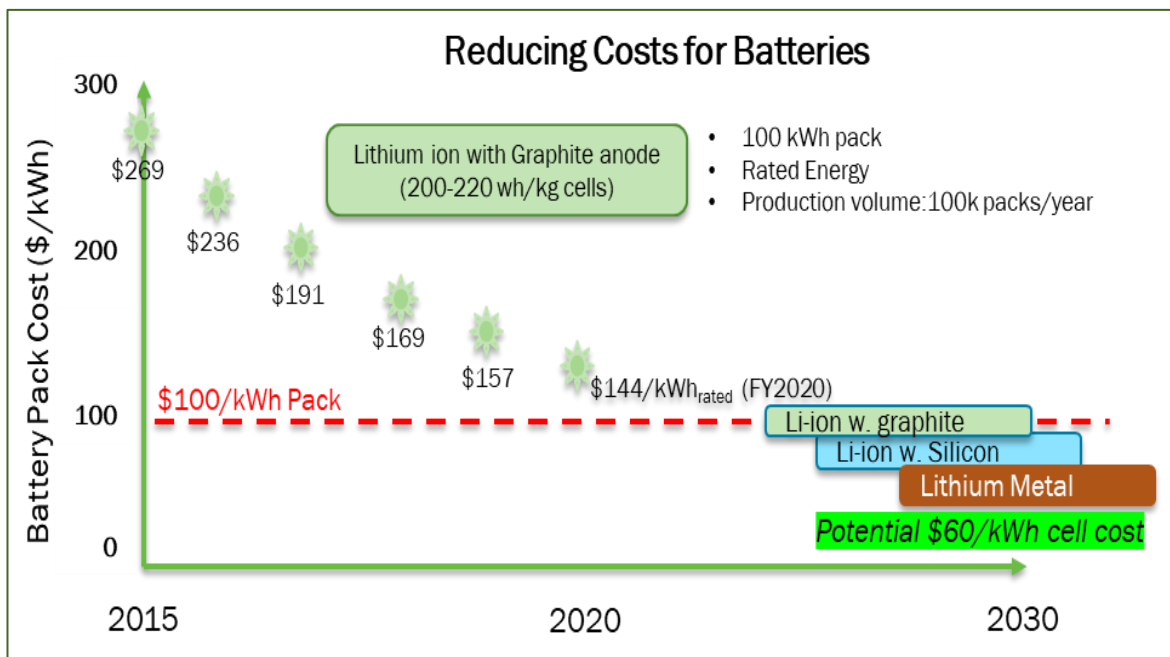
- Recycling and Sustainability
 - Cost to recycle is currently 5-15% of battery cost
 - Material shortage (Li, Co, and Ni)
 - Varying chemistries result in variable backend value

Partners

- Argonne National Laboratory
- National Renewable Energy Laboratory
- Oak Ridge National Laboratory
- University of California, San Diego
- Worcester Polytechnic Institute
- Michigan Technological University

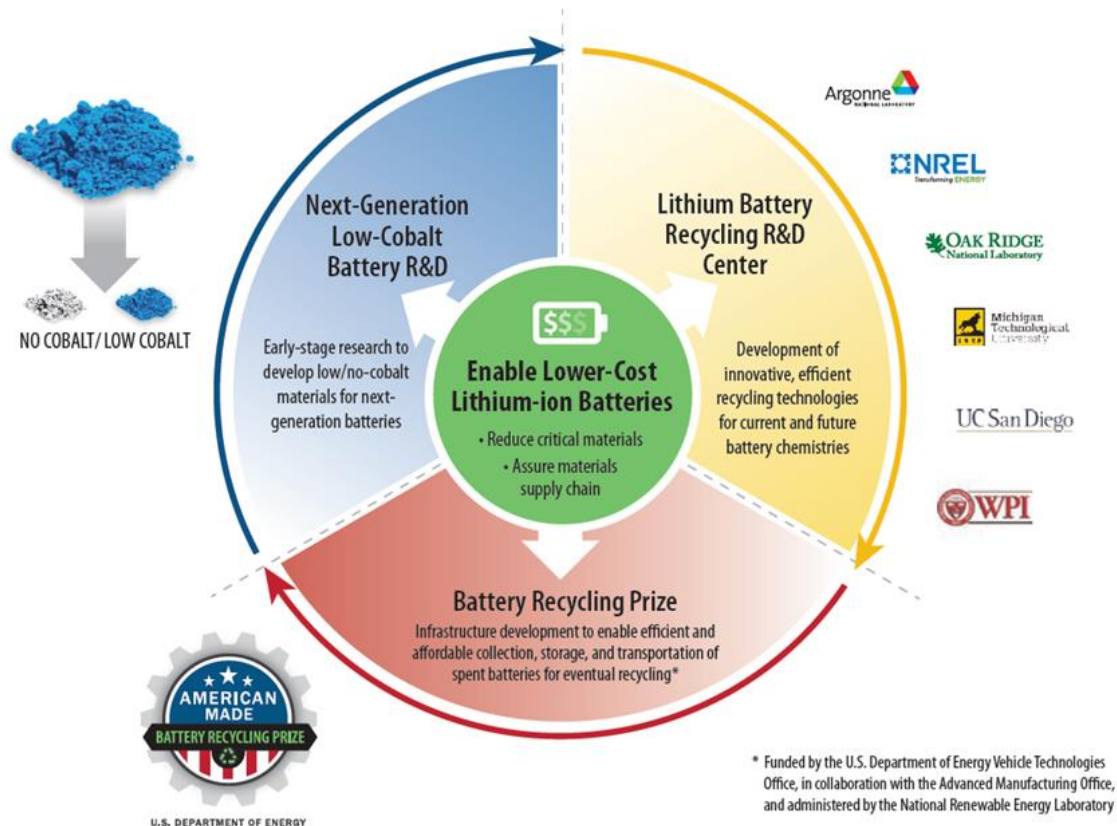
RELEVANCE

By 2025, reduce the cost of EV battery packs to less than \$100/kWh with technologies that **significantly reduce or eliminate the dependency on critical materials (such as cobalt) and utilize recycled material feedstocks.**



RELEVANCE

- Lower cost of batteries
- Enable lower environmental impacts
- Increase our country's energy security

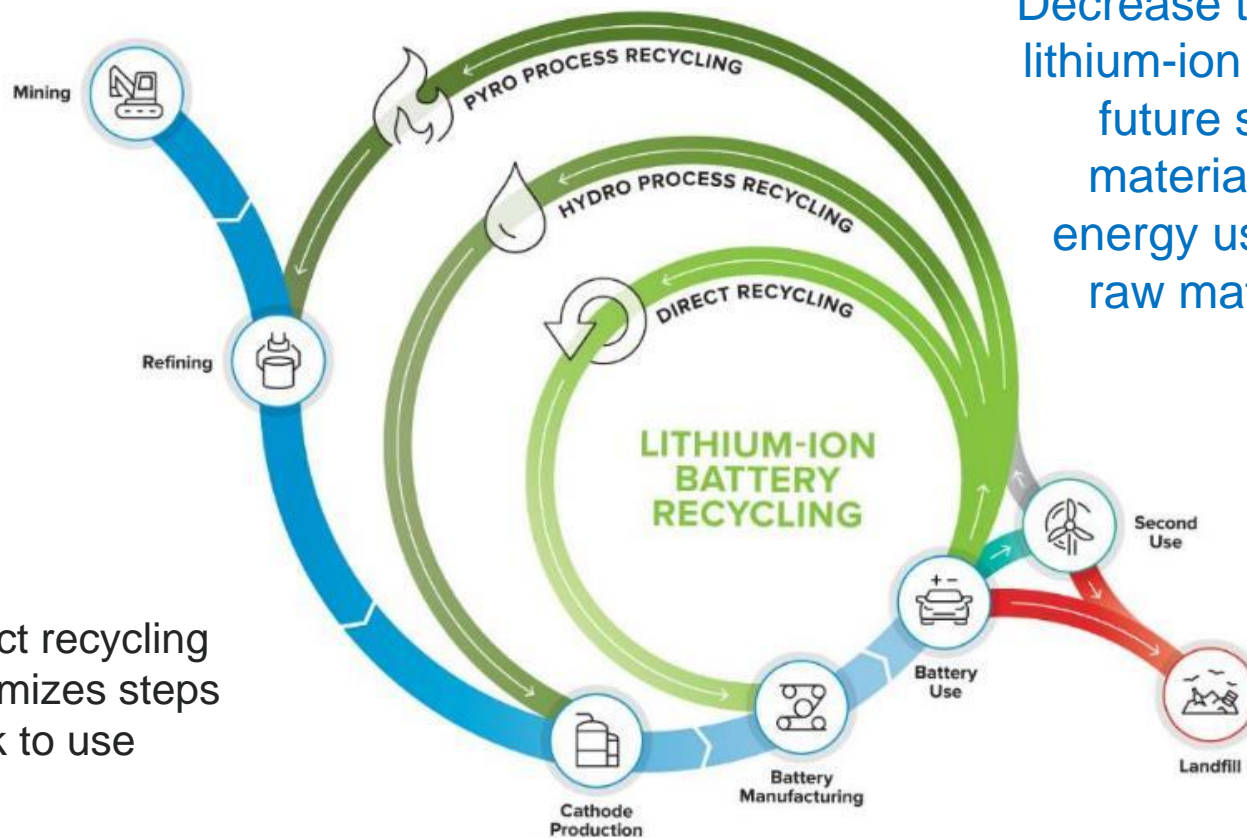


APPROACH

ReCell's Mission:

Decrease the cost of recycling lithium-ion batteries to ensure future supply of critical materials and decrease energy usage compared to raw material production

Direct recycling minimizes steps back to use



APPROACH

Year 1 – Bench scale testing:
Powder-to-Cell



Year 2 – Start to scale up
unit operations



Year 3 – Finish scale up and
show cell-to-cell recycling

- Binder Removal
- Cathode/ Cathode Separation
- Relithiation
- Cathode Upcycling
- Impurity Impact



**DIRECT
CATHODE
RECYCLING**

**OTHER
MATERIAL
RECOVERY**



- Cell Shredding
- Electrode Delamination
- Anode/ Cathode Separation
- Electrolyte Component Recovery

Cross Cutting Projects

- Cell Design for Rejuvenation



**DESIGN
FOR
SUSTAINABILITY**

**MODELING
AND
ANALYSIS**



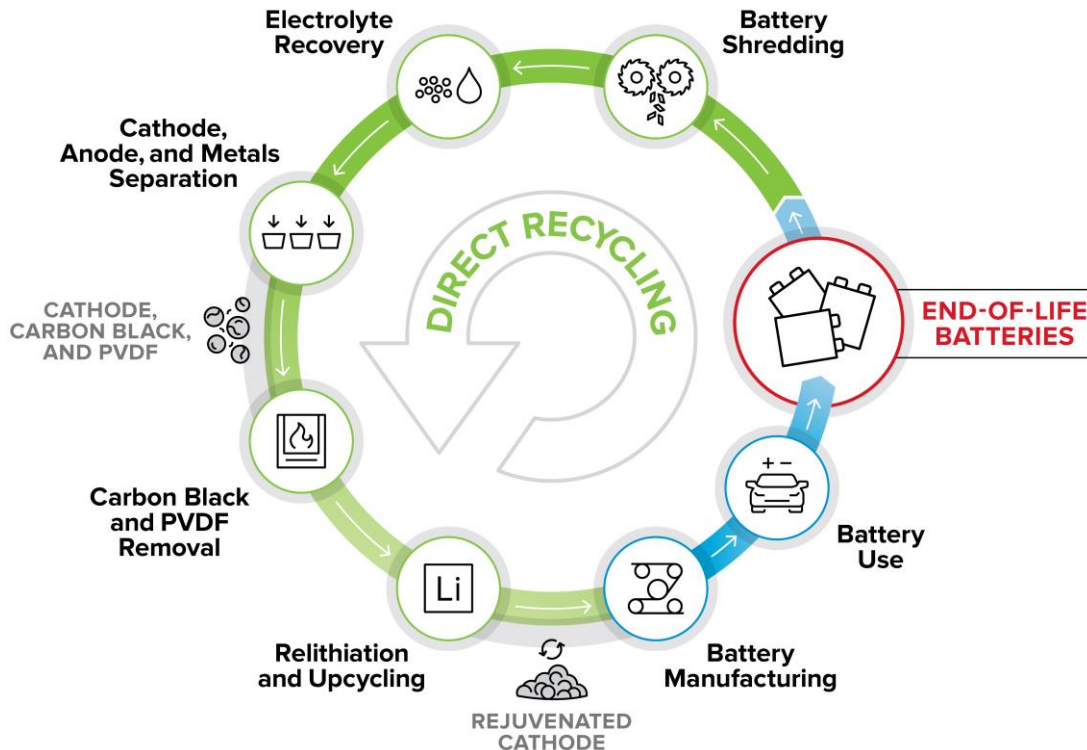
- EverBatt (TEA/LCA)
- LIBRA (Supply Chain Modeling)

Program does not include battery dismantling, transportation, or 2nd use

APPROACH

- Multiple processes investigated to mitigate risk
- Continual review of new project ideas
- End projects that are not showing promise in cost and performance
- These processes can benefit other recycling processes

Typical Direct Recycling Process Flow



MILESTONES

- | | | |
|---------|----------|---|
| FY20 Q3 | Complete | Down-select solvent(s) to separate black mass from current collector and optimize the process conditions to achieve >90% recovery of black mass |
| FY20 Q4 | Complete | Demonstrate recovery of anode and cathode powders using the new pilot scale froth column |
| FY21 Q1 | Complete | Preliminary report of sensitivity analysis of battery recycling in the LIBRA model focusing on outputs including the number of recycling plants built and the percentage of batteries recycled over time. |
| FY21 Q2 | Complete | Demonstrate 30% graphene yield from spent anode using a Taylor Vortex Reactor |
| FY21 Q3 | Ongoing | Final report on performance and cost modeling of directly recycled manufacturing scrap |
| FY21 Q4 | Ongoing | Provide preliminary cost analysis, yield, and efficiency on the separation-relithiation conditions on NMC spent electrodes via solvent-based dual process |

Each Individual project has its own milestones, though not listed here.

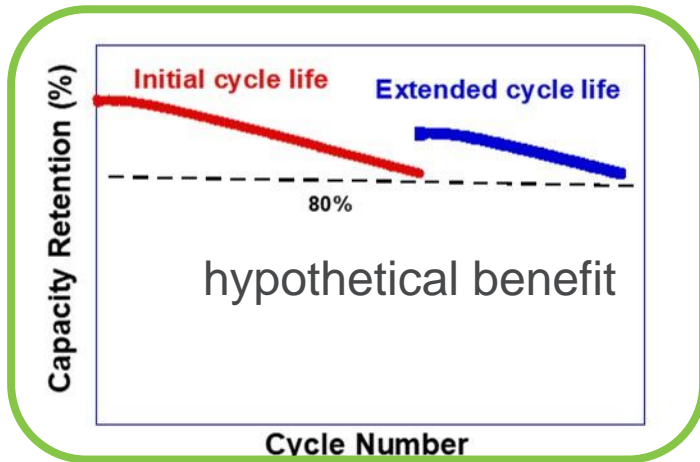
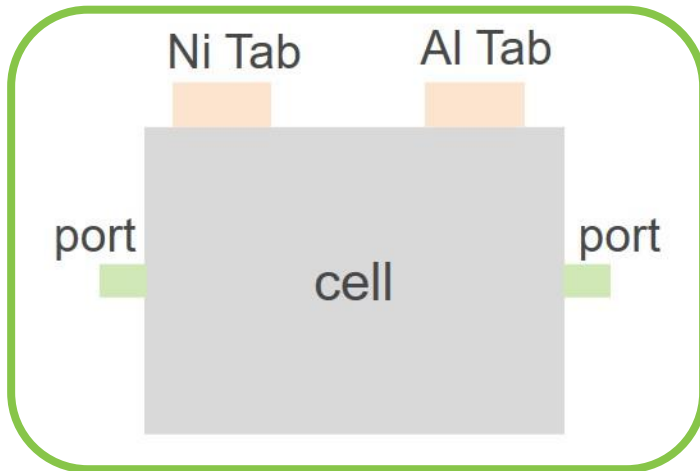
CELL DESIGN

Project Goal:

- Create cell designs to enable rejuvenation of a spent cell that trade minimal loss in energy-density performance for the ability to use cheaper, new recycling processes at end of life
- Relithiate spent cathodes to extend battery cycle life and reduce the number of batteries to be recycled

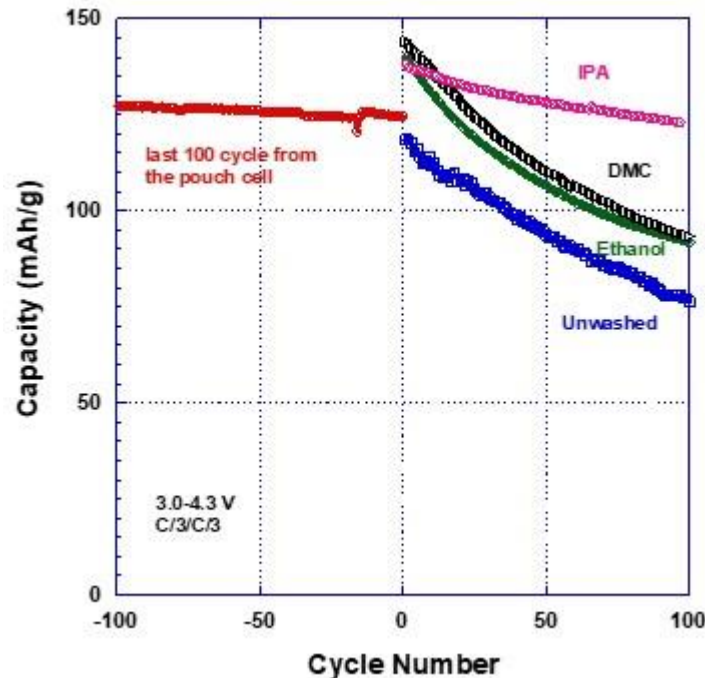
Project Description:

- Develop new cell designs with ports that allow us to flush off some SEI components and reduce cell polarization
- Identify optimal rejuvenation conditions, including solvents, flushing times and period
- Demonstrate extended cycle life in rejuvenated cells



INITIAL CAPACITY RECOVERY DEMONSTRATED IN COIN CELLS IN FY20

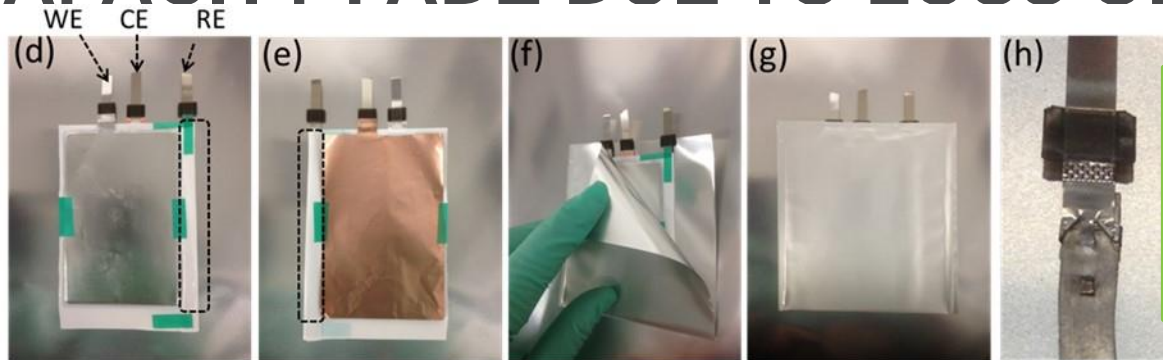
- Cycled pouch cells till 20% capacity fade
- Disassembled the spent cells
- Rinsed and dried the spent electrodes in glove box
- Assembled the rinsed electrodes into full coin cells
- Continued cycling the coin cells with same protocols



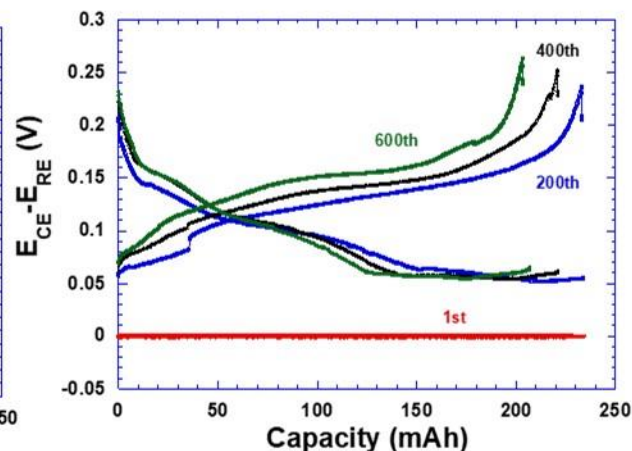
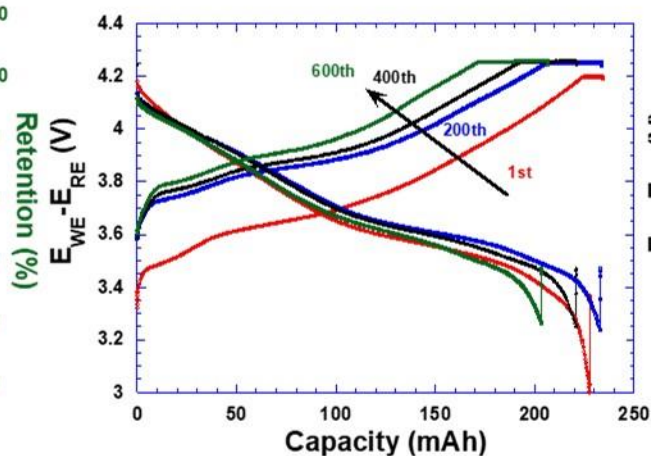
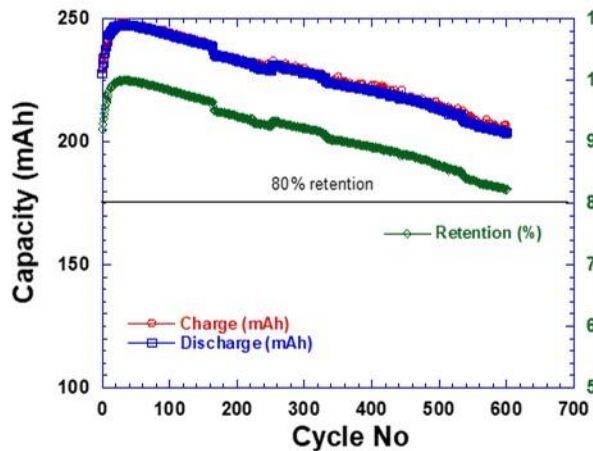
- Rinsing the spent electrode improved the capacity in the re-assembled cells albeit subsequent fast capacity degradation.
- The rinsed solution from the spent electrodes mainly consisted of POF_3 , $\text{C}_5\text{H}_{10}\text{O}_3$, $\text{C}_7\text{H}_{14}\text{O}_5$ and $\text{C}_8\text{H}_{14}\text{O}_6$ esters while $\text{C}_2\text{H}_6\text{O}$ was also found from the anode solution.

THREE-ELECTRODE POUCH CELLS VALIDATED

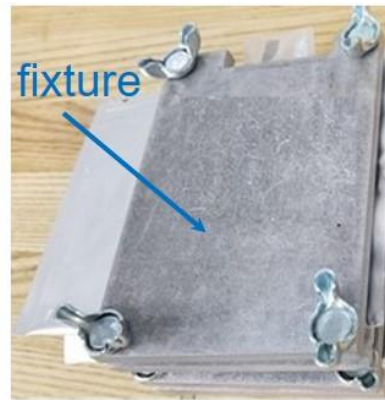
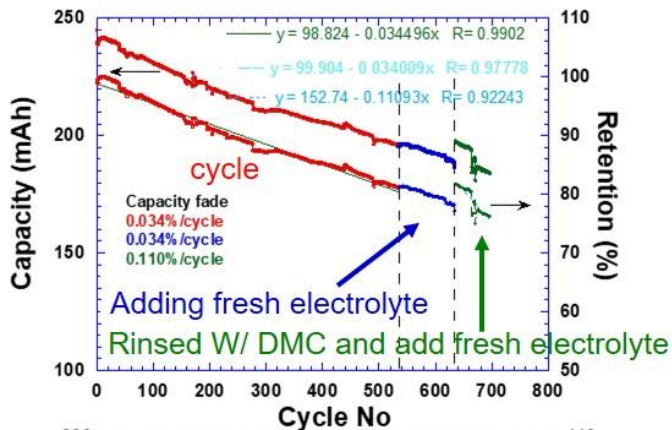
CAPACITY FADE DUE TO LOSS OF LI



- Good cycle life
- Capacity fade due to Li loss and resistance increase

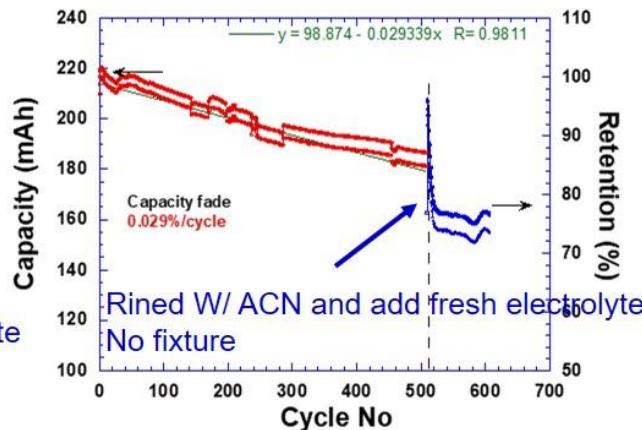
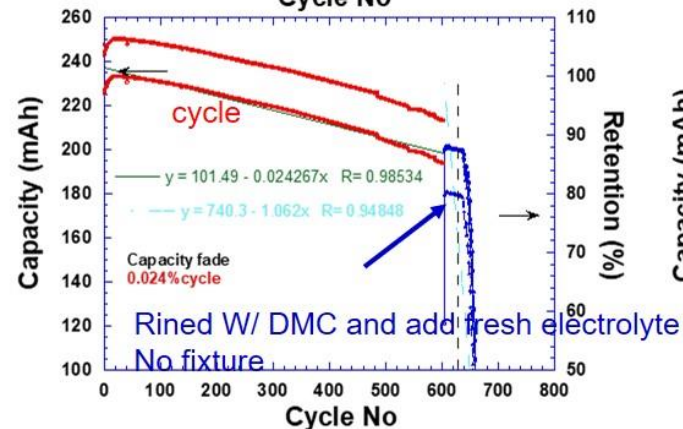


RINSING CELLS WITHIN TESTING FIXTURE ENABLE CAPACITY RECOVERY (SHORT TERM)



with the testing fixture:

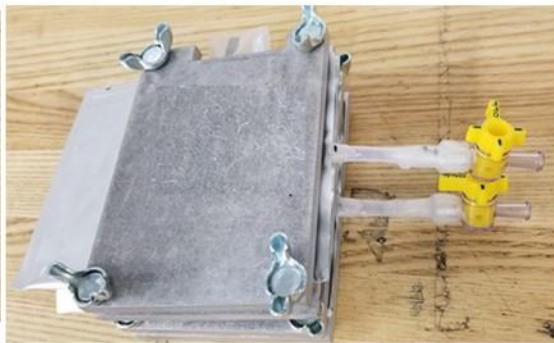
- Adding more electrolyte has insignificant effect in capacity recover
- Rinsing the cells and add fresh electrolyte showed short term capacity recovery but capacity faded quicker.



Without the testing fixture:

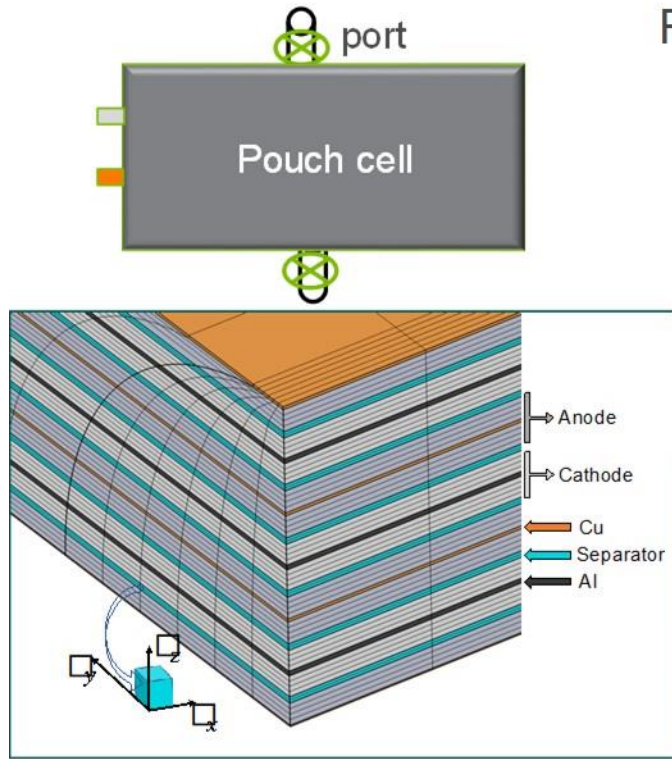
- adverse effect

FABRICATION OF POUCH CELLS WITH PORTS

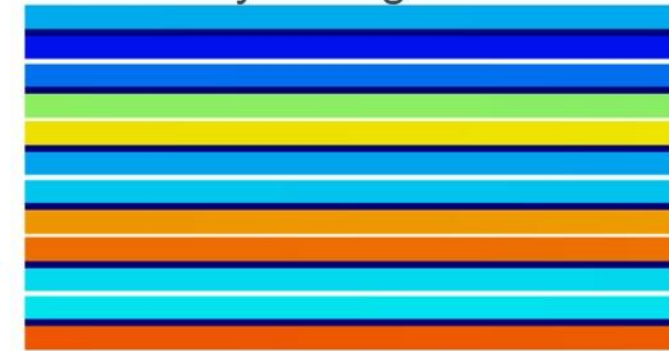


Challenge: difficult to maintain good sealing at the ports for long term, cells failed after a few days due to electrolyte loss

HIGHER FLUX THROUGH ANODE WHEN INCREASING GRAPHITE PARTICLE SIZE

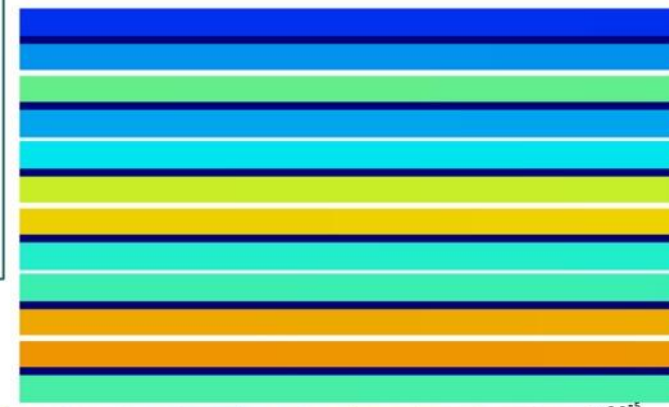


Flow velocity through cell mid-section



SLC1520P

5 10 15 20 25 30 35 40 45 50 $\times 10^{-5}$ m/s



SLC1506T

Higher flux through SLC 1520P than 1506T

REMAINING CHALLENGES AND FUTURE WORK

■ Key Challenges

- Leakage at the connection
- Low solubility of SEI in carbonate solvents
- Difficult to rinse cells under high pressure

■ Future Work

- Explore appropriate materials to seal the port with pouches
- Investigate liquid flow in electrodes with various microstructure

SUMMARY

- Validated loss of Li one main reason for capacity fade
- Demonstrated short term capacity recovery after cell rejuvenation in pouch cells
- Rinsing spent cells should be performed without releasing compression

We will complete the efforts in cell design for rejuvenation but will extend the efforts to design for sustainability.

- Reduce metal content in cell components
- Simplify pack design to access individual cells

RESPONSE TO REVIEWERS

Not reviewed last year

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**ADVANCED
BATTERY RECYCLING**

U.S. DEPARTMENT OF
ENERGY

Energy Efficiency &
Renewable Energy

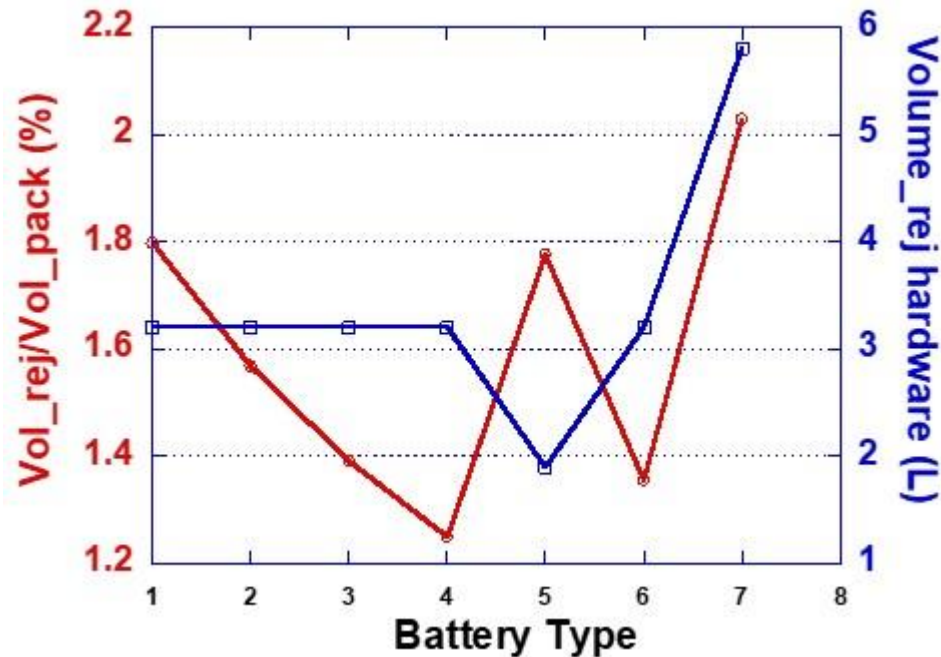
VEHICLE TECHNOLOGIES OFFICE

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Technical backup slides

REJUVENATION HARDWARE WOULD COMPROMISE ENERGY DENSITY BY 1-2%



- Estimation using BatPac
 - Assume: 2 tubes per cell, each 30 cm long & 0.4 cm OD (~4 mL each)
 - Assume: 2 valves per module (30 cells per module), ~60 mL each
 - Assume: 2 ganged tubing ports per module, ~15 mL each
 - Implied total of 400 mL per module if no wasted volume
 - Volume of the hardware for rejuvenation is 1.9 to 5.8 L.
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- The additional rejuvenation hardware would increase the pack volume by 1-2%.
- Courtesy of Andy Jansen